

IPCC Fourth Assessment Report: Climate Change 2007

Climate Change 2007: Working Group I: The Physical Science Basis

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2.10.2 Direct Global Warming Potentials



All GWPs depend on the AGWP for CO<sub>2</sub> (the denominator in the definition of the GWP). The AGWP of CO<sub>2</sub> again depends on the radiative efficiency for a small perturbation of CO<sub>2</sub> from the current level of about 380 ppm. The radiative efficiency per kilogram of CO<sub>2</sub> has been calculated using the same expression as for the CO<sub>2</sub> RF in Section 2.3.1, with an updated background CO<sub>2</sub> mixing ratio of 378 ppm. For a small perturbation from 378 ppm, the RF is 0.01413 W m<sup>-2</sup> ppm<sup>-1</sup> (8.7% lower than the TAR value). The CO<sub>2</sub> response function (see Table 2.14) is based on an updated version of the Bern carbon cycle model (Bern2.5CC; Joos et al. 2001), using a background CO<sub>2</sub> concentration of 378 ppm. The increased background concentration of CO<sub>2</sub> means that the airborne fraction of emitted CO<sub>2</sub> (Section 7.3) is enhanced, contributing to an increase in the AGWP for CO<sub>2</sub>. The AGWP values for CO<sub>2</sub> for 20, 100, and 500 year time horizons are 2.47 × 10<sup>-14</sup>, 8.69 × 10<sup>-14</sup>, and 28.6 × 10<sup>-14</sup> W m<sup>-2</sup> yr (kg CO<sub>2</sub>)<sup>-1</sup>, respectively. The uncertainty in the AGWP for CO<sub>2</sub> is estimated to be ±15%, with equal contributions from the CO<sub>2</sub> response function and the RF calculation.

Updated radiative efficiencies for well-mixed greenhouse gases are given in Table 2.14. Since the TAR, radiative efficiencies have been reviewed by Montzka et al. (2003) and Velders et al. (2005). Gohar et al. (2004) and Forster et al. (2005) investigated HFC compounds, with up to 40% differences from earlier published results. Based on a variety of radiative transfer codes, they found that uncertainties could be reduced to around 12% with well-constrained experiments. The HFCs studied were HFC-23, HFC-32, HFC-134a and HFC-227ea. Hurley et al. (2005) studied the infrared spectrum and RF of perfluoromethane (C<sub>2</sub>F<sub>6</sub>) and derived a 30% higher GWP value than given in the TAR. The RF calculations for the GWPs for CH<sub>4</sub>, N<sub>2</sub>O and halogen-containing well-mixed greenhouse gases employ the simplified formulas given in Ramaswamy et al. (2001; see Table 6.2 of the TAR). Table 2.14 gives GWP values for time horizons of 20, 100 and 500 years. The species in Table 2.14 are those for which either significant concentrations or large trends in concentrations have been observed or a clear potential for future emissions has been identified. The uncertainties of these direct GWPs are taken to be ±35% for the 5 to 95% (90%) confidence range.

Table 2.14. Lifetimes, radiative efficiencies and direct (except for CH<sub>4</sub>) GWPs relative to CO<sub>2</sub>. For ozone-depleting substances and their replacements, data are taken from IPCC/TEAP (2005) unless otherwise indicated.

Errata

Industrial Designation or Common Name (years)	Chemical Formula	Lifetime (years)	Radiative Efficiency (W m <sup>-2</sup> ppb <sup>-1</sup> )	Global Warming Potential for Given Time Horizon			
				SAR <sup>†</sup> (100-yr)	20-yr	100-yr	500-yr
Carbon dioxide	CO <sub>2</sub>	See below <sup>a</sup>	<sup>b</sup> 1.4x10 <sup>-5</sup>	1	1	1	1
Methane <sup>c</sup>	CH <sub>4</sub>	12 <sup>c</sup>	3.7x10 <sup>-4</sup>	21	72	25	7.6
Nitrous oxide	N <sub>2</sub> O	114	3.03x10 <sup>-3</sup>	310	289	298	153
<b>Substances controlled by the Montreal Protocol</b>							
CFC-11	CCl <sub>3</sub> F	45	0.25	3,800	6,730	4,750	1,620
CFC-12	CCl <sub>2</sub> F <sub>2</sub>	100	0.32	8,100	11,000	10,900	5,200
CFC-13	CClF <sub>3</sub>	640	0.25		10,800	14,400	16,400
CFC-113	CCl <sub>2</sub> FCClF <sub>2</sub>	85	0.3	4,800	6,540	6,130	2,700
CFC-114	CClF <sub>2</sub> CClF <sub>2</sub>	300	0.31		8,040	10,000	8,730
CFC-115	CClF <sub>2</sub> CF <sub>3</sub>	1,700	0.18		5,310	7,370	9,990
Halon-1301	CBrF <sub>3</sub>	65	0.32	5,400	8,480	7,140	2,760
Halon-1211	CBrClF <sub>2</sub>	16	0.3		4,750	1,890	575
Halon-2402	CBrF <sub>2</sub> CBrF <sub>2</sub>	20	0.33		3,680	1,640	503
Carbon tetrachloride	CCl <sub>4</sub>	26	0.13	1,400	2,700	1,400	435
Methyl bromide	CH <sub>3</sub> Br	0.7	0.01		17	5	1
Methyl chloroform	CH <sub>3</sub> CCl <sub>3</sub>	5	0.06		506	146	45
HCFC-22	CHClF <sub>2</sub>	12	0.2	1,500	5,160	1,810	549
HCFC-123	CHCl <sub>2</sub> CF <sub>3</sub>	1.3	0.14	90	273	77	24
HCFC-124	CHClF <sub>2</sub> CF <sub>3</sub>	5.8	0.22	470	2,070	609	185
HCFC-141b	CH <sub>3</sub> CCl <sub>2</sub> F	9.3	0.14		2,250	725	220
HCFC-142b	CH <sub>3</sub> CClF <sub>2</sub>	17.9	0.2	1,800	5,490	2,310	705
HCFC-225ca	CHCl <sub>2</sub> CF <sub>2</sub> CF <sub>3</sub>	1.9	0.2		429	122	37
HCFC-225cb	CHClF <sub>2</sub> CClF <sub>2</sub>	5.8	0.32		2,030	595	181
<b>Hydrofluorocarbons</b>							
HFC-23	CHF <sub>3</sub>	270	0.19	11,700	12,000	14,800	12,200

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HFC-32	CH <sub>2</sub> F <sub>2</sub>	4.9	0.11	650	2,330	675	205
HFC-125	CHF <sub>2</sub> CF <sub>3</sub>	29	0.23	2,800	6,350	3,500	1,100
HFC-134a	CH <sub>2</sub> FCF <sub>3</sub>	14	0.16	1,300	3,830	1,430	435
HFC-143a	CH <sub>3</sub> CF <sub>3</sub>	52	0.13	3,800	5,890	4,470	1,590
HFC-152a	CH <sub>3</sub> CHF <sub>2</sub>	1.4	0.09	140	437	124	38
HFC-227ea	CF <sub>3</sub> CHFCF <sub>3</sub>	34.2	0.26	2,900	5,310	3,220	1,040
HFC-236fa	CF <sub>3</sub> CH <sub>2</sub> CF <sub>3</sub>	240	0.28	6,300	8,100	9,810	7,660
HFC-245fa	CHF <sub>2</sub> CH <sub>2</sub> CF <sub>3</sub>	7.6	0.28		3,380	1030	314
HFC-365mfc	CH <sub>3</sub> CF <sub>2</sub> CH <sub>2</sub> CF <sub>3</sub>	8.6	0.21		2,520	794	241
HFC-43-10mee	CF <sub>3</sub> CHFCHFCF <sub>2</sub> CF <sub>3</sub>	15.9	0.4	1,300	4,140	1,640	500
<b>Perfluorinated compounds</b>							
Sulphur hexafluoride	SF <sub>6</sub>	3,200	0.52	23,900	16,300	22,800	32,600
Nitrogen trifluoride	NF <sub>3</sub>	740	0.21		12,300	17,200	20,700
PFC-14	CF <sub>4</sub>	50,000	0.10	6,500	5,210	7,390	11,200
PFC-116	C <sub>2</sub> F <sub>6</sub>	10,000	0.26	9,200	8,630	12,200	18,200

Table 2.14 (continued)

Industrial Designation or Common Name (years)	Chemical Formula	Lifetime (years)	Radiative Efficiency (W m <sup>-2</sup> ppb <sup>-1</sup> )	Global Warming Potential for Given Time Horizon			
				SAR <sup>a</sup> (100-yr)	20-yr	100-yr	500-yr
<b>Perfluorinated compounds (continued)</b>							
PFC-218		2,600	0.26	7,000	6,310	8,830	12,500
PFC-318		3,200	0.32	8,700	7,310	10,300	14,700
PFC-3-1-10		2,600	0.33	7,000	6,330	8,860	12,500
PFC-4-1-12		4,100	0.41		6,510	9,160	13,300
PFC-5-1-14		3,200	0.49	7,400	6,600	9,300	13,300
PFC-9-1-18		>1,000 <sup>d</sup>	0.56		>5,500	>7,500	>9,500
trifluoromethyl sulphur pentafluoride		800	0.57		13,200	17,700	21,200
<b>Fluorinated ethers</b>							
HFE-125		136	0.44		13,800	14,900	8,490
HFE-134		26	0.45		12,200	6,320	1,960
HFE-143a		4.3	0.27		2,630	756	230
HCFE-235da2		2.6	0.38		1,230	350	106
HFE-245cb2		5.1	0.32		2,440	708	215
HFE-245fa2		4.9	0.31		2,280	659	200
HFE-254cb2		2.6	0.28		1,260	359	109
HFE-347mcc3		5.2	0.34		1,980	575	175
HFE-347pcf2		7.1	0.25		1,900	580	175
HFE-356pcc3		0.33	0.93		386	110	33
HFE-449sl (HFE-7100)		3.8	0.31		1,040	297	90
HFE-569sf2 (HFE-7200)		0.77	0.3		207	59	18
HFE-43-10pccc124 (H-Galden 1040x)		6.3	1.37		6,320	1,870	569
HFE-236ca12 (HG-10)		12.1	0.66		8,000	2,800	860
HFE-338pcc13 (HG-01)		6.2	0.87		5,100	1,500	460
<b>Perfluoropolyethers</b>							
PFPME		800	0.65		7,620	10,300	12,400
<b>Hydrocarbons and other compounds – Direct Effects</b>							
Dimethylether		0.015	0.02		1	1	<<1
Methylene chloride		0.38	0.03		31	8.7	2.7
Methyl chloride		1.0	0.01		45	13	4

## Notes:

<sup>a</sup> The CO<sub>2</sub> response function used in this report is based on the revised version of the Bern Carbon cycle model used in [Chapter 10](#) of this report (Bern2.5CC; Joos et al. 2001) using a background CO<sub>2</sub> concentration value of 378 ppm. The decay of a pulse of CO<sub>2</sub> with time t is given by

$$a_0 + \sum_{i=1}^3 a_i \cdot e^{-t/\tau_i}$$

Where  $a_0 = 0.217$ ,  $a_1 = 0.259$ ,  $a_2 = 0.338$ ,  $a_3 = 0.186$ ,  $\tau_1 = 172.9$  years,  $\tau_2 = 18.51$  years, and  $\tau_3 = 1.186$  years.

<sup>b</sup> The radiative efficiency of CO<sub>2</sub> is calculated using the IPCC (1990) simplified expression as revised in the TAR, with an updated background concentration value of 378 ppm and a perturbation of +1 ppm (see [Section 2.10.2](#)).

<sup>c</sup> The perturbation lifetime for methane is 12 years as in the TAR (see also [Section 7.4](#)). The GWP for methane includes indirect effects from enhancements of ozone and stratospheric water vapour (see [Section 2.10.3.1](#)).

<sup>d</sup> Shine et al. (2005c), updated by the revised AGWP for CO<sub>2</sub>. The assumed lifetime of 1,000 years is a lower limit.

<sup>e</sup> Hurley et al. (2005)

<sup>f</sup> Robson et al. (2006)

<sup>g</sup> Young et al. (2006)

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